

PROTEINS

revised 30 Sept 2106

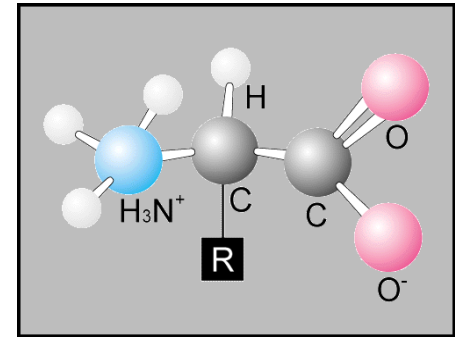
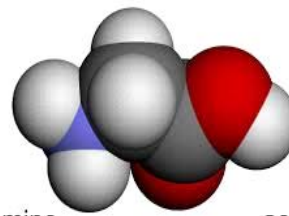
Campbell 6th: 71-80, 7th: 77-85, Sadava 8th: p42-49, Campbell 9th: 77-86, 10th: 75-83

Proteins: Once thought to be the foremost molecules of biology (therefore: *protein*)
functions:(p. 76)

- | | |
|-------------|---|
| enzymes | protein catalysts: amylase, pepsin, etc |
| defensive | antibodies, compliment |
| storage | ovalbumin, casein (both serve as AA source for development) |
| transport | hemoglobin, LDL, HDL |
| hormones | peptide hormones like insulin, etc |
| receptors | in nerves |
| contractile | actin and myosin in muscle |
| structure | silk, keratin, collagen |

amino group

acid group



Chemical definition

LEARN!: a linear polymer of amino acids connected by peptide bonds.

Review amino and carboxyl groups (p 78), both ionize at pH 7:
zwitterion [double charged particle]

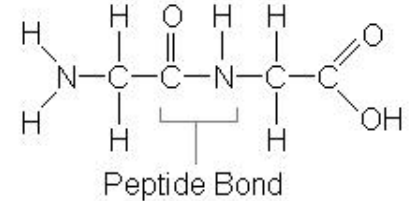
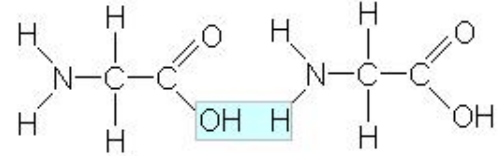
Has one asymmetric carbon (L in most proteins) (some weird microbial peptides: D.)

Properties of 20 AA depend on **side chain** (R group) (P.77)

- | | | |
|-------------------|------------------------------------|---------------|
| 9 are hydrocarbon | (hydrophobic) | glycine |
| 6 are polar | (hydrophilic) | cysteine |
| 5 are ionic | (pH dependent “-“ and “+” attract) | aspartic acid |
| | | lysine |

LEARN these 4 amino acids:

A molecule of water is removed from two glycine amino acids to form a peptide bond.



Proteins unidirectional, written N terminal to C terminus connected by peptide bonds (p 80)

PROTEIN STRUCTURE (LEARN): (p 80-81)

Folding of protein is spontaneous: **Self Assembly** (IMPORTANT!)

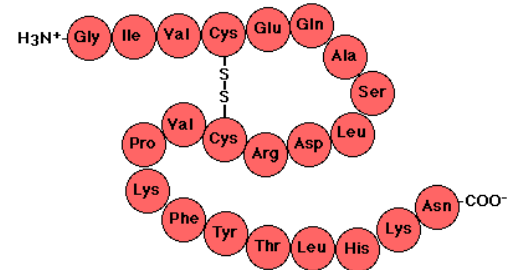
p 80 **primary:** Linear sequence (determines everything else) (mutation alters)

- secondary:** local interactions due to bonding between peptide H bonds, *not* side chains: highly predictable.
alpha helix a la Pauling and Corey, 1951, keratin, collagen
β pleated sheet: fibroin in silk, troughs and peaks
fibrous proteins have repeating structure, favor helix and pleated

[see excellent 3D demo at: <http://www.pdb.org/pdb/explore/jmol.do?structureId=5RSA&bionumber=1>]

p 81 **tertiary:** Depends on (non-repetitive) side chains, not easily predictable
Side chain H bonds, ionic bonds, hydrophobic bonds, disulfide
Disulfide bridges: rearrange for **permanent wave: 1: reduce, 2: set, 3: oxidize**
globular proteins rely more on tertiary

quaternary: multimeric proteins, above 50,000 MW same forces as as 3rd.
hemoglobin (a tetramer) is a classic example.



ENZYMES (“inside yeast”): **LEARN:** 2 word definition: Protein catalysts (p 153)
p 152 Illustrate profile of exergonic chemical reaction, role of **catalyst** (“down break”)
p. 132 Diagram an enzyme (**LEARN**):

- apoenzyme** (protein portion)
prosthetic group (required to assist apoenzyme)
cofactor: inorganic prosthetic group
coenzyme: organic prosthetic group (vitamins)
holoenzyme: active complex of apo- + prosthetic group
active site binds substrate, performs action

Lucy Chocolate factory: <https://www.youtube.com/watch?v=Jm1VEO9C4VQ>

Configuration is critical for enzyme activity.

Factors which affect configuration (p 156):
Inhibition of enzymes explain antiseptics, cooking, etc

pH (Affects charge on ionic side chains)
temperature affects tightness of folding
osmolarity: salt and sugar disrupt
disrupt S-S bonds (Heavy metals.)

